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(SECTION 1) PRODUCT IDENTIFICATION

Product Identifier

Product Name: Glycerine or Glycerol

Other means of Identification: Not Available

Relevant identified uses of the substance or mixture and uses advised against

Relevant Identified Uses:As solvent, humectant, plasticiser, emollient, sweetener; in the manufacture of nitroglycerol (explosive),

cosmetics, liquid soaps, liqueurs, confectioneries, blacking, printing and copying inks and lubricants. It is also used in the manufacture of elastic glues, lead oxide cements; to keep fabrics pliable; to preserve printing on cotton; for printing rollers; to keep frost from windshields; as antifreeze in automobiles, gas meters and hydraulic jacks, in shock absorber fluids. In fermentation nutrients in production of

antibiotics.

SECTION 2 HAZARD IDENTIFICATION

Classification of the substance or mixture

NON-HAZARDOUS CHEMICAL. NON-DANGEROUS GOODS. According to the WHS Regulations and the ADG Code.

CHEMWATCH HAZARD RATINGS

	Min	Max	
Flammability	1		0 = Minimum
Toxicity	0		1 = Low
Body Contact	0		2 = Moderate
Reactivity	1		3 = High
Chronic	0		4 = Extreme

Poinsons Schedule Not Applicable

LABEL ELEMENTS

GHS label elements: Not Applicable
Signal Word: Not Applicable

Hazard statement(s): Not Applicable

PRECAUTIONARY STATEMENT(S)

Prevention:Not ApplicableResponse:Not ApplicableStorage:Not ApplicableDisposal:Not Applicable

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SECTION 3

COMPOSITION/INFORMATION ON INGREDIENTS

Substances

See section below for composition of Mixtures

Mixtures

CAS No.	%[weight]	Name
56-81-5	>96	Glycerol

SECTION 4 FIRST AID MEASURES

Description of First Aid Measures

Eye Contact: If this product comes in contact with the eyes:

• Wash out immediately with fresh running water.

• Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the

eyelids by occasionally lifting the upper and lower lids.

• Seek medical attention without delay; if pain persists or recurs seek medical attention.

Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.

Skin Contact: If skin contact occurs:

• Immediately remove all contaminated clothing, including footwear.

• Flush skin and hair with running water (and soap if available).

• Seek medical attention in event of irritation.

Inhalation: • Lay patient down. Keep warm and rested.

• Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to

initiating first aid procedures.

• Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve

mask device, or pocket mask as trained. Perform CPR if necessary.

• Transport to hospital, or doctor, without delay.

Ingestion: • Immediately give a glass of water.

• First aid is not generally required. If in doubt, contact a Poisons Information Centre or a doctor.

Indication of any immediate medical attention and special treatment needed

Treat symptomatically.

SECTION 5 FIRE FIGHTING MEASURES

Extinguishing Media

- Water spray or fog.
- Foam.
- Dry chemical powder.
- BCF (where regulations permit).
- · Carbon dioxide.

Special hazards arising from the substrate or mixture

Fire Incompatibility:• Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result

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Advice for Fire Fighters

Fire Fighting:

- Alert Fire Brigade and tell them location and nature of hazard.
- Wear full body protective clothing with breathing apparatus.
- Prevent, by any means available, spillage from entering drains or water course.
- Use water delivered as a fine spray to control fire and cool adjacent area.
- · Avoid spraying water onto liquid pools.
- DO NOT approach containers suspected to be hot.
- Cool fire exposed containers with water spray from a protected location.
- If safe to do so, remove containers from path of fire.

Fire/Explosion Hazard:

- Combustible.
- Slight fire hazard when exposed to heat or flame.
- Heating may cause expansion or decomposition leading to violent rupture of containers.
- On combustion, may emit toxic fumes of carbon monoxide (CO).
- May emit acrid smoke.
- Mists containing combustible materials may be explosive

Combustion products include:

Carbon Dioxide (CO₂)

Acrolein

,

other pyrolysis products typical of burning organic material.

May emit poisonous fumes. May emit corrosive fumes.

HAZCHEM: Not Applicable

SECTION 6 ACCIDENTAL RELEASE MEASURES

Personal precautions, protective equipment and emergency procedures

See section 8

Environmental precautions

See section 12

Methods and material for containment and cleaning up

Minor Spills:

Slippery when spilt.

- · Remove all ignition sources
- Clean up all spills immediately.
- Avoid breathing vapours and contact with skin and eyes.
- Control personal contact with the substance, by using protective equipment.
- Contain and absorb spill with sand, earth, inert material or vermiculite. Wipe up.
- Place in a suitable, labelled container for waste disposal.

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Major Spills:

Slippery when spilt.

Moderate hazard.

- Clear area of personnel and move upwind.
- · Alert Fire Brigade and tell them location and nature of hazard.
- Wear breathing apparatus plus protective gloves.
- Prevent, by any means available, spillage from entering drains or water course.
- · No smoking, naked lights or ignition sources.
- · Increase ventilation.
- Stop leak if safe to do so.
- Contain spill with sand, earth or vermiculite.
- Collect recoverable product into labelled containers for recycling.
- Absorb remaining product with sand, earth or vermiculite.
- Collect solid residues and seal in labelled drums for disposal.
- Wash area and prevent runoff into drains.
- If contamination of drains or waterways occurs, advise emergency services.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

(SECTION 7)

HANDLING AND STORAGE

Precautions for Safe Handling

Safe Handling:

- DO NOT allow clothing wet with material to stay in contact with skin
- Avoid all personal contact, including inhalation.
- Wear protective clothing when risk of exposure occurs.
- Use in a well-ventilated area.
- Prevent concentration in hollows and sumps.
- DO NOT enter confined spaces until atmosphere has been checked.
- · Avoid smoking, naked lights or ignition sources.
- · Avoid contact with incompatible materials.
- When handling, DO NOT eat, drink or smoke.
- · Keep containers securely sealed when not in use.
- Avoid physical damage to containers.
- Always wash hands with soap and water after handling.
- Work clothes should be laundered separately.
- Use good occupational work practice.
- Observe manufacturer's storage and handling recommendations contained within this SDS.
- Atmosphere should be regularly checked against established exposure standards to ensure safe working conditions.

Other Information:

- Store in original containers.
- Keep containers securely sealed.
- No smoking, naked lights or ignition sources.
- Store in a cool, dry, well-ventilated area.
- Store away from incompatible materials and foodstuff containers.
- Protect containers against physical damage and check regularly for leaks.
- · Observe manufacturer's storage and handling recommendations contained within this SDS.

Conditions for safe storage, including any incompatibilities

Suitable Container:

- Metal can or drum
- Packaging as recommended by manufacturer.
- Check all containers are clearly labelled and free from leaks.

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Storage Incompatibility:

Glycerol:

- Reacts violently with strong oxidisers, acetic anhydride, alkali metal hydrides, calcium hypochlorite, calcium oxychloride, chlorine, chromic anhydride, chromium oxides, ethylene oxide, hydrogen peroxide, phosphorous triiodide, potassium chlorate, potassium permanganate, potassium peroxide, silver perchlorate, sodium hydride, sodium peroxide, sodium triiodide, sodium tetrahydroborate,is incompatible with strong acids, caustics, aliphatic amines, isocyanates, uranium fluoride
- Is able to polymerise above 145°C Avoid reaction with oxidising agents

SECTION 8 EXPOSURE CONTROLS/PERSONAL PROTECTION

CONTROL PARAMETERS

OCCUPATIONAL EXPOSURE LIMITS (OEL)

INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
Australia Exposure Standards	Glycerol	Glycerin mist	10 mg/m ³	Not Available	Not Available	Not Available

EMERGENCY LIMITS

Glycerol Glycerine (mist); (Glycerol; Glycerin) 45 mg/m³ 860 mg/m³ 2,500 mg/m³	Ingredient	Material name	TEEL-1	TEEL-2	TEEL-3
	Glycerol	Glycerine (mist); (Glycerol; Glycerin)	I Δ5 mσ/m³	860 mg/m ³) 500 mg/m ³

Ingredient	Original IDLH	Revised IDLH
Glycerol	Not Available	Not Available

Exposure Controls

Appropriate Engineering Controls:

Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection.

The basic types of engineering controls are:

Process controls which involve changing the way a job activity or process is done to reduce the risk. Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment. Ventilation can remove or dilute an air contaminant if designed properly. The design of a ventilation system must match the particular process and chemical or contaminant in use.

Employers may need to use multiple types of controls to prevent employee overexposure.

General exhaust is adequate under normal operating conditions. Local exhaust ventilation may be required in special circumstances. If risk of overexposure exists, wear approved respirator. Supplied-air type respirator may be required in special circumstances. Correct fit is essential to ensure adequate protection. Provide adequate ventilation in warehouses and enclosed storage areas. Air contaminants generated in the workplace possess varying "escape" velocities which, in turn, determine the "capture velocities" of fresh circulating air required to effectively remove the contaminant.

Type of Contaminant:	Air Speed:
Solvent, vapours, degreasing etc., evaporating from tank (in still air).	0.25-0.5 m/s (50- 100 f/min)
Aerosols, fumes from pouring operations, intermittent container filling, low speed conveyer transfers, welding, spray drift, plating acid fumes, pickling (released at low velocity into zone of active generation)	0.5-1 m/s (100- 200 f/min.)

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Direct spray, spray painting in shallow booths, drum filling, conveyer loading, crusher dusts, gas discharge (active generation into zone of rapid air motion)	1-2.5 m/s (200- 500 f/min.)
grinding, abrasive blasting, tumbling, high speed wheel generated dusts (released at high initial velocity into zone of very high rapid air motion)	2.5-10 m/s (500- 2000 f/min.)

Within each range the appropriate value depends on:

Lower end of the range	Upper end of the range
1: Room air currents minimal or favourable to capture	1: Disturbing room air currents
2: Contaminants of low toxicity or of nuisance value only.	2: Contaminants of high toxicity
3: Intermittent, low production.	3: High production, heavy use
4: Large hood or large air mass in motion	4: Small hood-local control only

Simple theory shows that air velocity falls rapidly with distance away from the opening of a simple extraction pipe. Velocity generally decreases with the square of distance from the extraction point (in simple cases). Therefore the air speed at the extraction point should be adjusted, accordingly, after reference to distance from the contaminating source. The air velocity at the extraction fan, for example, should be a minimum of 1-2 m/s (200-400 f/min) for extraction of solvents generated in a tank 2 meters distant from the extraction point. Other mechanical considerations, producing performance deficits within the extraction apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when extraction systems are installed or used.

Personal Protection:



PPE Gloves



PPE Goggles



PPE Mask



PPE Lab Coat

Eve and Face Protection:

- Safety glasses with side shields.
- · Chemical goggles.
- Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation lens should be removed in a clean environment only after workers have washed hands thoroughly. [CDC NIOSH Current Intelligence Bulletin 59], [AS/NZS 1336 or national equivalent]

Skin Protection:

Hands/Feet Protection:

See Hand protection below

• Wear chemical protective gloves, e.g. PVC.

• Wear safety footwear or safety gumboots, e.g. Rubber

The selection of suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer. Where the chemical is a preparation of several substances, the resistance of the glove material can not be calculated in advance and has therefore to be checked prior to the application.

The exact break through time for substances has to be obtained from the manufacturer of the protective gloves and has to be observed when making a final choice.

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Personal hygiene is a key element of effective hand care. Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried thoroughly. Application of a non-perfumed moisturizer is recommended.

Suitability and durability of glove type is dependent on usage. Important factors in the selection of gloves include:

- · frequency and duration of contact,
- · chemical resistance of glove material,
- · glove thickness and
- · dexterity

Select gloves tested to a relevant standard (e.g. Europe EN 374, US F739, AS/NZS 2161.1 or national equivalent).

- When prolonged or frequently repeated contact may occur, a glove with a protection class of 5 or higher (breakthrough time greater than 240 minutes according to EN 374, AS/NZS 2161.10.1 or national equivalent) is recommended.
- When only brief contact is expected, a glove with a protection class of 3 or higher (breakthrough time greater than 60 minutes according to EN 374, AS/NZS 2161.10.1 or national equivalent) is recommended.
- Some glove polymer types are less affected by movement and this should be taken into account when considering gloves for long-term use.
- · Contaminated gloves should be replaced.

For general applications, gloves with a thickness typically greater than 0.35 mm, are recommended. It should be emphasised that glove thickness is not necessarily a good predictor of glove resistance to a specific chemical, as the permeation efficiency of the glove will be dependent on the exact composition of the glove material. Therefore, glove selection should also be based on consideration of the task requirements and knowledge of breakthrough times.

Glove thickness may also vary depending on the glove manufacturer, the glove type and the glove model. Therefore, the manufacturers' technical data should always be taken into account to ensure selection of the most appropriate glove for the task.

Note: Depending on the activity being conducted, gloves of varying thickness may be required for specific tasks. For example:

- Thinner gloves (down to 0.1 mm or less) may be required where a high degree of manual dexterity is needed. However, these gloves are only likely to give short duration protection and would normally be just for single use applications, then disposed of.
- Thicker gloves (up to 3 mm or more) may be required where there is a mechanical (as well as a chemical) risk i.e. where there is abrasion or puncture potential

Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried thoroughly. Application of a non-perfumed moisturiser is

recommended.

Body protection: See Other protection below

Other protection: • Overalls.

- P.V.C. apron.
- · Barrier cream.
- Skin cleansing cream.
- Eye wash unit.

Thermal hazards: Not Available

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Recommended Material(s)

GLOVE SELECTION INDEX

Glove selection is based on a modified presentation of the:

"Forsberg Clothing Performance Index".

The effect(s) of the following substance(s) are taken into account in the computergenerated selection:

PALMERA G995B (GLYCERINE BP)

Material	СРІ
NATURAL RUBBER	A
NATURAL+NEOPRENE	A
NITRILE	A

^{*} CPI - Chemwatch Performance Index

A: Best Selection

B: Satisfactory; may degrade after 4 hours continuous immersion C: Poor to Dangerous Choice for other than short term immersion NOTE: As a series of factors will influence the actual performance of the glove, a final selection must be based on detailed observation. - * Where the glove is to be used on a short term, casual or infrequent basis, factors such as "feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

Respiratory Protection

Type A-P Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Where the concentration of gas/particulates in the breathing zone, approaches or exceeds the "Exposure Standard" (or ES), respiratory protection is required.

Degree of protection varies with both face-piece and Class of filter; the nature of protection varies with Type of filter.

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
up to 10 x ES	A-AUS P2	-	A-PAPR-AUS / Class 1 P2
up to 50 x ES	-	A-AUS / Class 1 P2	-
up to 100 x ES	-	A-2 P2	A-PAPR-2 P2^

^ - Full-face

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65°C)

Cartridge respirators should never be used for emergency ingress or in areas of unknown vapour concentrations or oxygen content. The wearer must be warned to leave the contaminated area immediately on detecting any odours through the respirator. The odour may indicate that the mask is not functioning properly, that the vapour concentration is too high, or that the mask is not properly fitted. Because of these limitations, only restricted use of cartridge respirators is considered appropriate.

(SECTION 9)

PHYSICAL/CHEMICAL PROPERTIES

Information on basic physical and chemical properties

Appearance: Oily, colourless, odourless liquid, with syrupy, sweet taste. Miscible with water and alcohol. Insoluble in

benzene, ether, chloroform, fixed and volatile oils. Absorbs water from the air. Also absorbs hydrogen

sulfide, hydrogen cyanide and sulfur dioxide.

Physical state:LiquidRelative Density (Water = 1):1.2-1.3 @ 20 COdour:Not AvailablePartition CoefficientNot Available

N-Octanol / Water:

Odour Threshold:Not AvailableAuto-Ignition Temperature (°C):370pH (as supplied):Not ApplicableDecomposition Temperature:280

Melting Point/Freezing Point 18 Viscosity (cSt): Not Available

(°C):

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Not Available



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Initial Boiling Point and

Boiling Range (°C):

Flash Point (°C):

160

160

290

Evaporation Rate: Non volatile Flammability: Not Available

Upper Explosive Limit (%): 19

opper Explosive Lillie (70).

Lower Explosive Limit (%): 3

Vapour Pressure (kPa): 0.003 @ 50 C
Solubility in Water: Miscible
Vapour Density (Air=1): 3.17

Molecular Weight (g/mol): 92.1

Taste: Not Available
Explosive Properties: Not Available
Oxidising Property: Not Available

Surface Tension (dyn/cm or mN/m):

Volatile Component (%vol): Nil @ 38 C

Gas Group: Not Available pH as a Solution (1%): 7 approx.

VOC g/L: Not Available

SECTION 10 STABILITY AND REACTIVITY

Reactivity: See section 7

Chemical Stability: • Unstable in the presence of incompatible materials

• Product is considered stable

Hazardous polymerisation will not occur

Possibility of Hazardous

Reaction:

See section 7

Conditions to Avoid:See section 7Incompatible Materials:See section 7Hazardous DecompositionSee section 5

Products:

SECTION 11 TOXICOLOGICAL INFORMATION

Information on Toxicological Effects

Inhaled: Not normally a hazard due to non-volatile nature of product

The material is not thought to produce adverse health effects or irritation of the respiratory tract (as classified by EC Directives using animal models). Nevertheless, good hygiene practice requires that exposure be kept to a minimum and that suitable control measures be used in an occupational setting.

Ingestion: Although ingestion is not thought to produce harmful effects (as classified under EC Directives), the

material may still be damaging to the health of the individual, following ingestion, especially where p

re-existing organ (e.g. liver, kidney) damage is evident.

Ilngestion of large quantities may cause nausea, diarrhoea and vomiting. Biokinetic studies indicate that a 70 kg adult should be able to metabolise and excrete over 2 grams of glycerol in an 8-hour workday.

Skin Contact: The material may accentuate any pre-existing dermatitis condition

Skin contact is not thought to have harmful health effects (as classified under EC Directives); the material may still produce health damage following entry through wounds, lesions or abrasions.

Open cuts, abraded or irritated skin should not be exposed to this material

Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.

formation provided on this document is presented in good faith and believe

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Eye:

Although the liquid is not thought to be an irritant (as classified by EC Directives), direct contact with the eye may produce transient discomfort characterised by tearing or conjunctival redness (as with

windburn).

Chronic:

Long-term exposure to the product is not thought to produce chronic effects adverse to the health (as classified by EC Directives using animal models); nevertheless exposure by all routes should be minimised as a matter of course.

PALMERA G995B (GLYCERINE BP):

TOXICITY	IRRITATION	
Intraperitoneal (Mouse) LD50: 8700 mg/kg ^[2]	Not Available	
Intraperitoneal (Rat) LD50: 4420 mg/kg ^[2]		
Intravenous (Mouse) LD50: 4250 mg/kg ^[2]		
Intravenous (Rat) LD50: 5566 mg/kg ^[2]		
Oral (Guinea pig) LD50: 7750 mg/kg ^[2]		
Oral (Mouse) LD50: 4090 mg/kg ^[2]		
Oral (Rat) LD50: 12600 mg/kg ^[2]		
Subcutaneous (Mouse) LD50: 91 mg/kg ^[2]		
Subcutaneous (Rat) LD50: 100 mg/kg ^[2]		

Glycerol:

TOXICITY	IRRITATION
Dermal (guinea pig) LD50: 54000 mg/kg ^[1]	Not Available
Oral (rat) LD50: >20-<39800 mg/kg ^[1]	

Legend: 1. Value obtained from Europe ECHA Registered Substances - Acute toxicity 2.* Value obtained from manufacturer's SDS. Unless otherwise specified data extracted from RTECS - Register of Toxic Effect of chemical Substances

PALMERA G995B:

(GLYCERINE BP) & GLYCEROL:

Asthma-like symptoms may continue for months or even years after exposure to the material ceases. This may be due to a non-allergenic condition known as reactive airways dysfunction syndrome (RADS) which can occur following exposure to high levels of highly irritating compound. Key criteria for the diagnosis of RADS include the absence of preceding respiratory disease, in a non-atopic individual, with abrupt onset of persistent asthma-like symptoms within minutes to hours of a documented exposure to the irritant. A reversible airflow pattern, on spirometry, with the presence of moderate to severe bronchial hyperreactivity on methacholine challenge testing and the lack of minimal lymphocytic inflammation, without eosinophilia, have also been included in the criteria for diagnosis of RADS. RADS (or asthma) following an irritating inhalation is an infrequent disorder with rates related to the concentration of and duration of exposure to the irritating substance. Industrial bronchitis, on the other hand, is a disorder that occurs as result of exposure due to high concentrations of irritating substance (often particulate in nature) and is completely reversible after exposure ceases. The disorder is characterised by dyspnea, cough and mucus production.

PALMERA G995B (GLYCERINE BP) & GLYCEROL: At very high concentrations, evidence predicts that glycerol may cause tremor, irritation of the skin, eyes, digestive tract and airway. Otherwise it is of low toxicity. There is no significant evidence to suggest that it causes cancer, genetic, reproductive or developmental toxicity.

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Acute Toxicity:	×	Carcinogenicity:	5
Skin Irritation/Corrosion:	5	Reproductivity:	5
Serious Eye Damage/Irritation:	5	STOT - Single Exposure:	5
Respiratory or Skin Sensitisation:	5	STOT - Repeated Exposure:	5
Mutagenicity:	5	Aspiration Hazard:	150

Legend: X – Data available but does not fill the criteria for classification

💚 – Data available to make classification

🔯 – Data Not Available to make classification

SECTION 12 ECOLOGICAL INFORMATION

Toxicity

Ingredient	Endpoint	Test Duration (hr)	Species	Value	Source
Glycerol	LC50	96	Fish	>11mg/L	2
Glycerol	EC50	96	Algae or other aquatic plants	77712.039mg/L	3
Glycerol	EC20	24	Crustacea	>500mg/L	1
Legend:	Extracted from 1. IUCLID Toxicity Data 2. Europe ECHA Registered Substances - Ecotoxicological Information - Aquatic Toxicity 3. EPIWIN Suite V3.12- Aquatic Toxicity Data (Estimated) 4. US EPA, Ecotox database - Aquatic Toxicity Data 5. ECETOC Aquatic Hazard Assessment Data 6. NITE (Japan) - Bioconcentration Data 7. METI (Japan) - Bioconcentration Data 8. Vendor Data				

For Glycerol: Log Kow: -2.66 to -2.47, Atmospheric Fate: Glycerol is broken down in the air by hydroxyl radicals the half-life for this process is 6.8 hours. However, only a negligible amount of the substance will move to the atmospheric compartment. Terrestrial Fate: Only a negligible amount of glycerol will move into the soil compartment, if released into the environment. Aquatic Fate: Glycerol is considered to be readily biodegradable in the aquatic environment. Pre-adapted microorganisms can break glycerol down rapidly in oxygenated/low oxygen waters. The substance is not expected to react with water. When released to water, 100% of the substance will remain in the water compartment only negligible amounts will be distributed to sediment. DO NOT discharge into sewer or waterways.

Persistence and Degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
Glycerol	LOW	LOW

Bioaccumulative Potential

Ingredient	Bioaccumulation
Glycerol	LOW (LogKOW = -1.76)

Mobility in Soil

Ingredient	Mobility	
Glycerol	HIGH (KOC = 1)	

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DISPOSAL CONSIDERATIONS

Waste Treatment Methods Product/Packaging Disposal:

(SECTION 14) TRANSPORT INFORMATION

Label Required

Marine Pollutant:

NO

HAZCHEM:

No Applicable

Land transport (ADG): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Air transport (ICAO-IATA/DGR): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Sea transport (IMDG-Code / GGV See): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Transport in bulk according to Anex II of MARPOL and the IBC code

Source	Product Name	Pollution Category	Ship Type
IMO MARPOL (Anex II - List of Noxious Liquid Substances Carried in Bulk	Glycerine	Z	Not Applicable

(SECTION 15)

REGULATORY INFORMATION

Safety, health and environmental regulations / legislation specific for the substance or mixture **GLYCEROL(56-81-5) IS FOUND ON THE FOLLOWING REGULATORY LISTS**

Australia Exposure Standards

Australia Inventory of Chemical Substances (AICS)

•	,
National Inventory	
Australia - AICS	Υ
Canada - DSL	Υ
Canada - NDSL	N (Glycerol)
China - IECSC	Υ
Europe - EINEC / ELINCS / NLP	Υ
Japan - ENCS	Υ
Korea - KECI	Υ
New Zealand - NZloC	Υ
Philippines - PICCS	Υ
USA - TSCA	Υ
Legend:	Y = All ingredients are on the inventory N = Not determine or one or more ingredients are not on the inventory and are not exempted from listing (see specific ingredients in brackets)

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SECTION 16 OTHER INFORMATION

Other Information:

Ingredients with multiple CAS number

Name	CAS No.
Glycerol	56-81-5, 29796-42-7, 30049-52-6, 37228-54-9, 75398-78-6, 78630-16-7, 8013-25-0

Definitions and abbreviations

PC-TWA: Permissible Concentration-Time Weighted Average PC-STEL: Permissible Concentration-Short Term Exposure Limit

IARC: International Agency for Research on Cancer

ACGIH: American Conference of Governmental Industrial Hygienists

STEL: Short Term Exposure Limit

TEEL: Temporary Emergency Exposure Limit,

IDLH: Immediately Dangerous to Life or Health Concentrations

OSF: Odour Safety Factor

NOAEL: No Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level

TLV: Threshold Limit Value LOD: Limit Of Detection OTV: Odour Threshold Value BCF: BioConcentration Factors BEI: Biological Exposure Index

Reason for Revision: To bring to date

END OF SDS