

Issue Date: 29/04/2004

Revision Date: 09/11/2017

Supersedes Date: 01/07/2016

Version Number: 04

## SAFETY DATA SHEET



HEALTH	3
FIRE	2
REACTIVITY	0
PERSONAL PROTECTION	G

Product Code: AJA809-500ML

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### 1. Product Identification

**GHS Product Identifier** FORMALDEHYDE SOLUTION

**Recommended use of the chemical and restrictions on use** Disinfectant, germicide, fungicide, insecticide, manufacture of organic chemicals, explosives, rubber, resins and dyes, photography, tanning, fabric treatment, chemical analysis and laboratory reagent.

**Other Names** Oxymethylene, Formic aldehyde, Methanal, Formalin

### 2. Hazard Identification

**GHS classification of the substance/mixture** Carcinogenicity: Category 1  
Acute Toxicity - Dermal: Category 3  
Eye Damage/Irritation: Category 1  
Germ Cell Mutagenicity: Category 2  
Acute Toxicity - Inhalation: Category 3  
Acute Toxicity - Oral: Category 4  
Sensitization - Respiratory: Category 1  
Skin Corrosion/Irritation: Category 1B  
Sensitization - Skin: Category 1

**Signal Word (s)** DANGER

**Hazard Statement (s)** H302 Harmful if swallowed.  
H311 Toxic in contact with skin.  
H314 Causes severe skin burns and eye damage.  
H317 May cause an allergic skin reaction.  
H331 Toxic if inhaled.  
H341 Suspected of causing genetic defects.  
H350 May cause cancer.  
H370 Causes damage to organs.

**Pictogram (s)** Corrosion, Skull and crossbones, Exclamation mark, Health hazard



**Precautionary statement – Prevention** P201 Obtain special instructions before use.  
P202 Do not handle until all safety precautions have been read and understood.  
P260 Do not breathe dust/fume/gas/mist/vapours/spray.  
P264 Wash thoroughly after handling.  
P270 Do not eat, drink or smoke when using this product.  
P271 Use only outdoors or in a well-ventilated area.  
P272 Contaminated work clothing should not be allowed out of the workplace.

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### Precautionary statement – Response

P280 Wear protective gloves/protective clothing/eye protection/face protection.

Swallowed

P301+P330+P331 IF SWALLOWED: rinse mouth. Do NOT induce vomiting.

P310 Immediately call a POISON CENTER or doctor/physician.

Skin

P303+P361+P353 IF ON SKIN (or hair): Remove/Take off immediately all contaminated clothing. Rinse skin with water/shower.

P363 Wash contaminated clothing before reuse.

P310 Immediately call a POISON CENTER or doctor/physician.

Inhaled

P304+P340 IF INHALED: Remove victim to fresh air and keep at rest in a position comfortable for breathing.

P310 Immediately call a POISON CENTER or doctor/physician.

Eyes

P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

P310 Immediately call a POISON CENTER or doctor/physician.

P307+P311 IF exposed: Call a POISON CENTER or doctor/physician.

P403+P233 Store in a well-ventilated place. Keep container tightly closed.

### Precautionary statement – Storage

P405 Store locked up.

### Precautionary statement – Disposal

P501 Dispose of contents/container to an approved waste disposal plant.

## 3. Composition/Information on Ingredients

Chemical Liquid

### Characterization

Information on Stabilised with methanol.

### Composition

### Ingredients

Name	CAS
Water	7732-18-5
Formaldehyde	50-00-0
Methanol	67-56-1

## 4. First Aid Measures

### Inhalation

Remove victim from exposure - avoid becoming a casualty. Remove contaminated clothing and loosen remaining clothing. Allow patient to assume most comfortable position and keep warm. Keep at rest until fully recovered. If breathing laboured and patient cyanotic (blue), ensure airways are clear and have qualified person give oxygen through a face mask. If breathing has stopped apply artificial respiration at once. In the event of cardiac arrest, apply external cardiac massage. Seek urgent medical assistance.

### Ingestion

Rinse mouth thoroughly with water immediately, repeat until all traces of product have been removed. Give water to drink. DO NOT INDUCE VOMITING. Seek immediate medical advice.

### Skin

Wash affected areas with copious quantities of water immediately. Remove contaminated clothing and wash before re-use. For skin burns, immediately flood burnt area with plenty of water. Cover with a clean, dry dressing. Seek urgent medical assistance.

### Eye contact

Immediately irrigate with copious quantity of water for at least 15 minutes. Eyelids to be held open. Seek medical attention.

### First Aid Facilities

Maintain eyewash fountain and safety shower in work area.

### Advice to Doctor

Treat symptomatically based on judgement of doctor and individual reactions of the patient.

### Other Information

For advice, contact a Poisons Information Centre (Phone eg Australia 13 1126;) or a doctor.

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### Hazards from Combustion Products

May liberate toxic fumes in fire including formic acid, methanol, carbon monoxide and carbon dioxide.

### Specific Methods

Combustible liquid

Small fire: Use foam, dry chemical, CO<sub>2</sub> or water spray.

Large fire: Use foam, fog or water spray. Do not use water jets.

If safe to do so, move undamaged containers from fire area. Cool containers with flooding quantities of water until well after fire is out. Avoid getting water inside containers.

Alcohol resistant foam is preferred however fine water spray can be used.

### Specific hazards arising from the chemical

May be ignited by heat, sparks or flame. Vapours can form explosive mixtures with air. Vapours may travel to source of ignition and flash back. Containers may explode when heated. Vapours from runoff may create an explosion hazard. Fire will produce irritating, poisonous and/or corrosive gases.

### Hazchem Code

•2X

### Precautions in connection with Fire

Wear SCBA, fully-encapsulating, gas-tight suit and structural firefighting uniform when handling leaking or damaged containers and equipment. SCBA and chemical splash suits will offer limited protection for brief exposure provided there is no risk of ignition.

## 6. Accidental Release Measures

### Spills & Disposal

ELIMINATE all ignition sources (no smoking, flares, sparks or flames) within at least 25m - All equipment used when handling the product must be earthed. Do not touch or walk through spilled material. Stop leak if safe to do so - Prevent entry into waterways, drains or confined areas. Vapour-suppressing foam may be used to control vapours - Water spray may be used to knock down or divert vapour clouds. Absorb with earth, sand or other non-combustible material. Use clean, non-sparking tools to collect absorbed material and place it into loosely-covered metal or plastic containers for later disposal. SEEK EXPERT ADVICE ON HANDLING AND DISPOSAL.

### Personal

### Precautions

Evacuate the area of all non-essential personnel. Avoid inhalation, contact with skin, eyes and clothing.

### Personal Protection

Wear protective clothing specified for normal operations (see Section 8)

### Clean-up Methods - Small Spillages

Absorb or contain liquid with sand, earth or spill control material. Shovel up using non sparking tools and place in a labelled, sealable container for subsequent safe disposal. Put leaking containers in a labelled drum or overdrum.

### Clean-up Methods - Large Spillages

Seek expert advice on handling and disposal.

## 7. Handling and Storage

### Precautions for Safe Handling

Avoid generation of vapours/aerosols. Do not breathe vapour. Avoid contact with eyes, skin and clothing. Avoid prolonged or repeated exposure. Work under hood.

### Conditions for safe storage, including any

Store in cool place and out of direct sunlight. Store away from sources of heat or ignition. Store in well ventilated area. Store away from oxidising agents, acids, alkalis, metal salts and foodstuff. Keep containers closed at all times - check regularly for leaks.

### incompatibilities

### Corrosiveness

Corrosive to carbon steel and gray and ductile cast iron at 20 °C, due to the presence of formic acid. Not corrosive, at 20 °C, to most common metals, such as stainless steel, aluminium, high silicon cast iron, nickel and nickel-base alloys, naval brass, admiralty brass, naval bronze, tantalum, titanium and zirconium.

### Storage Regulations

Refer Australian Standard AS 3780-1994 'The storage and handling of corrosive substances'.

### Recommended Materials

Most plastics, such as Teflon and other fluorocarbons, acrylonitrile-butadiene-styrene (ABS), nylon 66, chlorinated polyvinyl chloride (CPVC), polyvinyl chloride (PVC), polyethylene and polyethylene; and elastomers, such as Viton, Chemraz, Kalrez and other fluorocarbons, ethylene propylene, butyl rubber, nitrile rubber (NBR), neoprene and low density polyethylene.

### Unsuitable Materials

Plastics, such as nylon 6, acrylic fibre (Orlon) and polystyrene (90); and elastomers, such as polyurethane, chloroprene, soft rubber, and isoprene.

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### 8. Exposure Controls/Personal Information

Occupational exposure limit values	Name	STEL		TWA		Footnote
		mg/m <sup>3</sup>	ppm	mg/m <sup>3</sup>	ppm	
	Formaldehyde	2.5	2	1.2	1	
	Methanol	328	250	262	200	

#### Other Exposure Information

A time weighted average (TWA) has been established for formaldehyde (Safe Work Australia) of 1.2 mg/m<sup>3</sup>, (1 ppm) and for methyl alcohol of 262 mg/m<sup>3</sup>, (200 ppm). The corresponding STEL level for formaldehyde is 2.5 mg/m<sup>3</sup>, (2 ppm) and for methyl alcohol is 328 mg/m<sup>3</sup> (250 ppm).

The STEL (Short Term Exposure Limit) is an exposure value that should not be exceeded for more than 15 minutes and should not be repeated for more than 4 times per day. There should be at least 60 minutes between successive exposures at the STEL. The exposure value at the TWA is the average airborne concentration of a particular substance when calculated over a normal 8 hour working day for a 5 day working week. Note: Absorption through skin may be a significant route of exposure for methyl alcohol.

Note: Sensitiser (for formaldehyde). Known to act as a sensitiser. - Safe Work Australia. Sensitiser notice: Some substances can cause a specific immune response in some people. Such substances are called sensitisers and the development of a specific immune response is termed 'sensitisation'. Exposure to a sensitiser, once sensitisation has occurred, may manifest itself as a skin rash or inflammation or as an asthmatic condition, and in some individuals this reaction can be extremely severe.

#### Appropriate engineering controls

In industrial situations maintain the concentrations values below the TWA. This may be achieved by process modification, use of local exhaust ventilation, capturing substances at the source, or other methods.

#### Respiratory Protection

Where ventilation is not adequate, respiratory protection may be required. Avoid breathing vapours or mists. Select and use respirators in accordance with AS 1716 - Respiratory Protective Devices and be selected in accordance with AS 1715 - Selection, Use and Maintenance of Respiratory Protective Devices. When mists or vapours exceed the exposure standards then the use of the following is recommended: Approved respirator with organic vapour and dust/mist filters. Filter capacity and respirator type depends on exposure levels.

#### Eye Protection

The use of a face shield, chemical goggles or safety glasses with side shield protection as appropriate. Must comply with Australian Standards AS 1337 and be selected and used in accordance with AS 1336.

#### Hand Protection

Hand protection should comply with AS 2161, Occupational protective gloves - Selection, use and maintenance. Recommendation: Excellent: Vinyl gloves. Good: NR latex, nitrile and neoprene. Avoid skin contact when removing gloves from hands, do not touch the gloves outer surface. Dispose of gloves as hazardous waste.

#### Personal Protective Equipment

Final choice of personal protective equipment will depend on individual circumstances and/or according to risk assessments undertaken.

#### Footwear

Safety boots in industrial situations is advisory, foot protection should comply with AS 2210, Occupational protective footwear - Guide to selection, care and use.

#### Body Protection

Clean impervious clothing should be worn, preferably with an apron for extra protection. Clothing for protection against chemicals should comply with AS 3765 Clothing for Protection Against Hazardous Chemicals.

#### Hygiene Measures

Always wash hands before smoking, eating or using the toilet. Wash contaminated clothing and other protective equipment before storing or re-using.

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### 9. Physical and Chemical Properties

<b>Form</b>	Liquid
<b>Appearance</b>	Colourless liquid; may become cloudy on standing.
<b>Odour</b>	Pungent, suffocating odour.
<b>Melting Point</b>	< -15 °C
<b>Freezing Point</b>	Varies with formaldehyde and methanol concentration. -13 to -11 °C (20% formaldehyde); approximately -16 °C (25% formaldehyde). Not available for more concentrated formaldehyde solutions or formaldehyde/methanol water solutions.
<b>Boiling Point</b>	100 °C
<b>Solubility in Water</b>	Miscible.
<b>Solubility in Organic Solvents</b>	Soluble in all proportions in alcohols, such as ethanol and methanol, and acetone.
<b>Specific Gravity</b>	1.09 @ 20 °C
<b>pH</b>	2.5 - 4.0
<b>Vapour Pressure</b>	Partial pressure of formaldehyde: 0.173 kPa (1.3 mm Hg) at 20 °C (37% formaldehyde); the presence of methanol increases the partial pressure of formaldehyde.
<b>Vapour Density (Air=1)</b>	1.04 (formaldehyde gas)
<b>Evaporation Rate</b>	The evaporation rate is expected to be low at normal temperatures.
<b>Odour Threshold</b>	0.027 - 1.9 ppm.
<b>Viscosity</b>	5.62 mPa.s (5.62 centipoises) at 25 °C (37% formaldehyde, 7% methanol) (calculated)
<b>Partition Coefficient: n-octanol/water</b>	Log P(oct) = 0.35 (experimental) (formaldehyde).
<b>Flash Point</b>	56 °C (closed cup); > 62 °C (Open Cup)
<b>Flammability</b>	Combustible liquid.
<b>Auto-Ignition Temperature</b>	~300 °C
<b>Flammable Limits - Lower</b>	7%
<b>Flammable Limits - Upper</b>	73%
<b>Molecular Weight</b>	30.03 (formaldehyde).
<b>Dynamic Viscosity</b>	5.62 mPa.s (5.62 centipoises) at 25 °C (calculated).
<b>Saturated Vapour Concentration</b>	1350-1700 ppm (0.135-0.17%) at 20 °C (36-37% formaldehyde) (calculated); the presence of methanol will increase the SVC of formaldehyde.
<b>Other Information</b>	Refractive index: 1.3746 @ 20 °C

### 10. Stability and Reactivity

<b>Chemical Stability</b>	Stable. Stabilised with methanol.
<b>Conditions to Avoid</b>	Open flames, heat, hot surfaces, sparks and other ignition sources.
<b>Incompatible Materials</b>	Strong oxidizing agents (e.g. hydrogen peroxide, potassium permanganate); strong bases (e.g. alkalis, such as sodium hydroxide); phenol; acrylonitrile; strong acids (e.g. sulfuric acid or acetic anhydride); performic acid; hydrochloric acid; aniline and perchloric acid; magnesium carbonate hydroxide; urea, isocyanates, anhydrides or oxides, polymerisation initiators (e.g. alkali metals), nitrogen oxides.
<b>Hazardous</b>	Formic acid, methanol, carbon monoxide and carbon dioxide.



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### Decomposition

### Products

### Possibility of

### hazardous reactions

Strong oxidizing agents (e.g. hydrogen peroxide, potassium permanganate) - may react violently, with the risk of fire and explosion; strong bases (e.g. alkalis, such as sodium hydroxide) - reaction produces flammable hydrogen gas, which may ignite. This reaction may lead to possible pressurization in closed containers, which may rupture. Phenol - runaway reactions have occurred during production of phenol-formaldehyde resins. Acrylonitrile - a violently exothermic and runaway reaction may result from the reaction between acrylonitrile and formaldehyde (as paraformaldehyde or trioxane) in the presence of strong acids (e.g. sulfuric acid or acetic anhydride). Performic acid - formaldehyde reacts violently with 90% performic acid. Hydrochloric acid - form highly toxic bis(chloromethyl)ether. Aniline and perchloric acid - aniline treated with perchloric acid, then formaldehyde gives a resinous condensation product, which burns with explosive violence. Magnesium carbonate hydroxide - reaction may release carbon dioxide gas, which may rupture closed containers. Urea, isocyanates, anhydrides, or oxides - may react vigorously or violently. Explosive with air in a vaporous/gaseous state when heated.

### Hazardous

### Polymerization

If unstabilised, formaldehyde solutions polymerise to paraformaldehyde. Polymerization is not hazardous. Methanol: polymerisation inhibitor.

## 11. Toxicological Information

### Toxicology Information

This substance should be treated with great care.

### Acute Toxicity - Oral

LD50 (rat): >200 mg/kg (Formaldehyde).

### Ingestion

Toxic if swallowed. Ingestion causes immediate irritation of the mouth, throat and stomach resulting in nausea. In extreme cases swallowing can result in vomiting, diarrhoea, abdominal pain, convulsions, chemical burns, loss of consciousness, collapse and possible death. Risk of perforation in the

### Inhalation

oesophagus and stomach. Systemic effects: narcosis and blindness. Toxic! Irreversible damage possible. Inhalation may lead to the formation of oedemas in the respiratory tract. Vapour is irritating to mucous membranes and the respiratory tract. Inhalation can result in headache, dizziness and possible nausea.

### Skin

Toxic in contact with skin. Corrosive to skin - may cause hardening or cracking of the skin, burns and dermatitis. Repeated or prolonged skin contact may lead to allergic contact dermatitis. A skin sensitizer. A component of this material (methanol) can be absorbed through the skin, however symptoms of poisoning via this route are unlikely because of low absorption. Danger of skin absorption. Irreversible damage is possible.

### Eye

Corrosive to eyes. Severe irritant to the eye. Vapour may cause inflammation of the eyelids. Contact can cause corneal burns. Contamination of the eyes can result in permanent injury.

### Skin Sensitisation

Known to act as a sensitizer.

### Carcinogenicity

Formaldehyde [50-00-0]: Group 2: The agent is probably carcinogenic to humans. Safe Work Australia Probable human carcinogens are those substances for which there is sufficient evidence to provide a strong presumption that human exposure might result in the development of cancer. This evidence is generally based on appropriate long term animal studies, limited epidemiological evidence or other relevant information.

Formaldehyde [50-00-0] is evaluated in the IARC Monographs (Vol. 88; in preparation) as Group 1: Carcinogenic to humans.

For addition information see IARC publication:

<http://monographs.iarc.fr/ENG/Monographs/vol100F/mono100F-29.pdf>

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### Reproductive Toxicity

Formaldehyde [resp], human: one study suggests a slight percentage increase in spontaneous abortion and subtle neurobehavioral abnormalities, animal-decreased sperm motility, reduced fetal and maternal weight.

### Chronic Effects

Repeated or prolonged skin contact may cause chronic dermatitis. Harmful: possible risk of irreversible effects through inhalation, in contact with skin and if swallowed. Chronic exposure to methanol from skin contact, inhalation and/or swallowing at concentrations greater than 1000 ppm can result in permanent blindness and central nervous system effects. Some long term animal test data suggests a carcinogenic potential for formaldehyde contained in this solution. This was found to occur at levels, which caused chronic tissue irritation and was well above the exposure standard. These particular data are not considered relevant to normal use because these high concentrations would not be voluntarily tolerated by humans, but do emphasise the need for care in handling.

### Serious eye damage/irritation

(Rabbit): Severe irritation

### Mutagenicity

Formaldehyde [50-00-0]: DNA damage system-human: fibroblast 100 mmol/l.

### Skin

(Rabbit): Severe irritation

### corrosion/irritation

## 12. Ecological Information

### Ecological Information

The following statements refer to individual components of the preparation:

### Persistence and degradability

Abiotic degradation: Rapid degradation. (air, formaldehyde)  
Biologic degradation: Biodegradation: 97.4 % /5 d (Formaldehyde). Readily biodegradable.  
COD: 1.06 g/g (Formaldehyde); TOD: 1.068 g/g (Formaldehyde)

### Mobility

Distribution: log p(o/w): 0.00 (Formaldehyde).

### Bioaccumulative Potential

No bioaccumulation is to be expected (log P(o/w) <1).

### Biological Properties

Toxic for aquatic organisms. Protoplasmic toxin. Caustic even in diluted form. Disinfectant effect. Toxic effect on fish and plankton. Sludge decomposition impaired or not possible even in diluted concentration. Endangers drinking-water supplies if allowed to enter soil and/or waters in large quantities.

### Environmental Protection

Do not allow to enter waters, waste water, or soil!

### Acute Toxicity - Fish

LC50 (*P.promelas*): 24 mg/l /96 h (Formaldehyde);  
LC50 (*Br.rerio*): 41 mg/l /96 h (Formaldehyde).

### Acute Toxicity - Daphnia

Daphnia *magna* EC50: ~2 mg/l /48 h (Formaldehyde).

### Acute Toxicity - Algae

Maximum permissible toxic concentration: Algal toxicity: *Sc.quadricauda* IC5: 2.5 mg/l /8 d (Formaldehyde).

### Acute Toxicity - Bacteria

Photobacterium *phosphoreum* EC50: 8.5 mg/l /30 min (Formaldehyde).  
Bacterial toxicity: *M.aeruginosa* EC5: 0.39 mg/l /8 d (Formaldehyde).

## 13. Disposal Considerations

### Disposal

Whatever cannot be saved for recovery or recycling should be disposed of according to relevant local, state and federal government regulations.

### Considerations

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### 14. Transport Information

<b>Transport Information</b>	Dangerous goods of Class 8 (Corrosive) are incompatible in a placard load with any of the following: Class 1, Class 4.3, Class 5, Class 6, if the Class 6 dangerous goods are cyanides and the Class 8 dangerous goods are acids, Class 7; and are incompatible with food and food packaging in any quantity.
<b>U.N. Number</b>	2209
<b>UN proper shipping name</b>	FORMALDEHYDE SOLUTION
<b>Transport hazard class(es)</b>	8
<b>Hazchem Code</b>	•2X
<b>Packaging Method</b>	3.8.8RT7,RT8
<b>Packing Group</b>	III
<b>EPG Number</b>	8A1
<b>IERG Number</b>	19

### 15. Regulatory Information

<b>Regulatory Information</b>	Listed in the Australian Inventory of Chemical Substances (AICS).
<b>Poisons Schedule</b>	S6

### 16. Other Information

<b>Literature References</b>	<p>'Standard for the Uniform Scheduling of Medicines and Poisons No. 6', Commonwealth of Australia, February 2015.</p> <p>Lewis, Richard J. Sr. 'Hawley's Condensed Chemical Dictionary 13th. Ed.', Rev., John Wiley and Sons, Inc., NY, 1997.</p> <p>National Road Transport Commission, 'Australian Code for the Transport of Dangerous Goods by Road and Rail 7th. Ed.', 2007.</p> <p>Safe Work Australia, 'National Code of Practice for the Preparation of Safety Data Sheets for Hazardous Chemicals', 2011.</p> <p>Standards Australia, 'SAA/SNZ HB 76:2010 Dangerous Goods - Initial Emergency Response Guide', Standards Australia/Standards New Zealand, 2010.</p> <p>Safe Work Australia, 'Approved Criteria for Classifying Hazardous Substances [NOHSC:1008 (2004)]'.</p> <p>Safe Work Australia, 'Hazardous Substances Information System, 2005'.</p> <p>Safe Work Australia, 'National Code of Practice for the Labelling of Safe Work Hazardous Substances (2011)'.</p> <p>Safe Work Australia, 'National Exposure Standards for Atmospheric Contaminants in the Occupational Environment [NOHSC:1003(1995)]'.</p>
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**Empirical Formula &** HCOH

**Structural Formula**



**Reason for Revision:** To comply with GHS Regulation

## END OF SDS